PLANNER & TRACKER FOR RECOVERY ANNUAL **TEACHING PLAN (ATP)**



7 GRADE

Helping teachers and learners to catch up with learning losses, master new content and acquire skills for the future.





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2021 - 2023

• Please note that a Natural Sciences structured learning programme that includes daily lesson plans and classroom resources is available for download from www.nect.org.za

- This is a zero-rated website, so there are no data costs for downloads.
- This document can be used independently of the structured learning programme.



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Planner & Tracker for Recovery ATP Natural Sciences



Grade 7 Term 2 2021 - 2023

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Introduction

Dear Natural Sciences Teachers,

The COVID-19 Pandemic has left us with an enormous challenge in education. As we return to 'normal schooling', we all have to work smarter and harder to ensure that our system recovers.

This document is designed to help you achieve this. By systematically working through this plan, we are confident that you can address the loss of teaching and learning time, and bring your learners to the level where they need to be in terms of NS.

We thank you in advance for the commitment, dedication and hard work that is required of you. You are truly building our nation.

With very best wishes for the term ahead, The DBE / NECT Recovery ATP Trackers Team

Overview

Please continue to keep the following key principles in mind throughout the recovery journey:

- The development of **Science Process Skills** is key to the teaching and learning of the subject. Focussing on these skills is critical.
- Learners should be given as many opportunities as possible to write regularly and read for meaning, in Natural Science, in order to develop language skills as well. Due to learning losses, as a result of the Covid pandemic, it is the responsibility of every educator to develop these literacy skills.
- It is very important to give learners a sense of how science applies to their daily lives, and
 of the value that science adds to their lives. Hold a brief discussion on this point when
 introducing a new topic, and invite learners to contribute their ideas on the uses and value that
 this topic has.
- At the end of every topic, come back to the topic overview, and **reflect on what has been learnt and taught**. In particular, it is important to note your challenges and ideas for future improvement, so that you can improve your teaching the next year.
- At the core of all scientific activities is the need to ask questions. These questions help us seek answers through observation and experimental design. The results of these questions should raise more questions. It is this natural curiosity that all teachers, and especially science teachers, should be encouraging in their classrooms. Encourage curiosity and questions that investigate, inquire and probe.
- **Build a solid conceptual foundation** for learners. A **conceptual chain** for the phase is provided at the start of this document. It is important for all NS teachers to work cohesively to ensure that learners are equipped with a solid understanding of the required concepts, by the time they leave the phase.
- Using the **CONCEPTUAL CHAIN** provided, **work together** as a department to:
 - a. Check that all concepts for the phase are covered in your school's recovery plan.
 - b. Check for overlaps across the grades.
 - c. **Identify the weak links in the conceptual chain** points where learners struggle and may be the source of misconceptions or common errors.
 - d. Decide how to **emphasise critical concepts from previous grades** especially where topics have moved from a different grade in the revised ATP.

Senior Phase Conceptual Chain: Grade 7



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Senior Phase Conceptual Chain: Grade 8



Senior Phase Conceptual Chain: Grade 9



Grade 7 Term 2 Natural Sciences - Planner & Tracker for Recovery ATP

The Recovery ATP for Natural Sciences has the **same content as in CAPS**, however, this content has been arranged as follows for Term 2:

Some topics from Grade 6 have been included/recovered:

1. Solids, liquids and gases	(0,5 weesk)
2. Solutions as a special mixture	(1 week)
3. Dissolving	(1 week)
Some topics remain :	
1. Properties of Materials	(reduced in time to 1,5 weeks)
2. Separating mixtures	(2 weeks)
3. Acids, bases and neutrals	(2 weeks)

Some topics have been **removed completely**:

1. Introduction to the Periodic Table of Elements

Directions on how to cover all required topics are provided in the Tracker that follows.

Amendments To The Programme Of Assessment

- The Programme of Assessment is aligned to the Revised Section 4 of CAPS.
- Both formal and informal assessment should continue as normal.
- Recording of the informal assessment is left to the discretion of the teacher.
- The 2022 formal assessment tasks for Grade 7 are as follows:

	TERM 1	TERM 2	Term 4	TERM 4
Practical Task/Investigation/Projects	20 marks	20 marks	30 marks	-
Test	60 marks	80 marks	60 marks	80 marks

A sample Assessment Test and Memorandum for Grade 7 Term 2 is included in this document.

Notes:

- Column 1 shows the time allocation per topic.
- Column 2 shows the Recovery ATP requirements for Grade 7 Term 2.
- **Column 3** explains any **changes** that have been made to the teaching plan.
- Column 4 shows where in the NECT lesson plans this is covered.
- Column 5 shows where in the approved textbooks this is covered.
- Finally, if, for any reason, the **Term 2 teaching time** for NS **is reduced**, please

ensure that the **KEY CONCEPTS** listed below each table are thoroughly covered.

Кеу То	Approved Textbook Abbreviations:
PLAT	Platinum Natural Sciences Grade 7 Maskew Miller Longman
SFA	Solutions for All Natural Sciences Grade 7 MacMillan
ох	Oxford Successful Natural Sciences Grade 7 Oxford University Press
SO	Spot On Natural Sciences Grade 7 Pearson
тс	Top Class Natural Sciences Grade 7 Shuter and Shooter
SIBB	Sasol Inzalo Bk B Natural Sciences Grade 7 Sasol
SbS	Step-by-Step Natural Sciences Grade 7 Van Schaik
VA	Via Afrika Natural Sciences Grade 7 Via Afrika
PEL	Pelican Natural Sciences Grade 7 Global MBD Africa

DATE COMPLETED									
OVED	59 – 61	54 - 57	56 - 60	89 - 93	58 - 63	48 - 49	32	51 - 53	104 - 117
APPR TEXTE	S&M Gr6	VIVA Gr6	PLAT Gr6	SFA Gr6	DbD Gr6	OX Gr6	SO Gr6	TC Gr6	SIBB Gr6
NECT LESSON PLANS: LESSONS	<u>Gr6 Term 2 Lesson Plans</u>	Lesson 1A: Arrangement of particles							
NOTES	This topic	recovered	trom Grade 6 Term 2						
DBE RECOVERY ATP REQUIREMENTS	Solids, liquids and	gases 1. Arrangement of	particles						
TIME ALLOCATION	Week 1	C,D WEEKS							

Scaling down

If the Term 2 teaching time is reduced, ensure that learners have a thorough understanding of the following key content and concepts:

Solids, liquids and gases

- Identify the 3 states of matter in everyday life.
- Explain how the distance between molecules helps identify what state the matter is in.
- Describe the different arrangements of particles in solids, liquids and gases.
- Illustrate the arrangement of particles in solids, liquids and gases.

DATE COMPLETED										
OVED OOKS	57 - 61	58 – 65	65 - 70	57 – 63	72 – 79	81 – 88	101 – 119	70–78	147 - 167	
APPR	OX Gr7	VA Gr7	PLAT Gr7	SO Gr7	SbS Gr7	PEL Gr7	SFA Gr7	TC Gr7	SIBB Gr7	
NECT LESSON PLANS: LESSONS	Gr 7 Term 2 Lesson Plans Lesson 1A: Physical properties of materials Lesson 1B: Boiling and melting points Lesson 1C: Heat and electrical conductivity Lesson 2A: Factors to consider when choosing materials								-	
NOTES	This topic	has been reduced to	1,5 weeks							
DBE RECOVERY ATP REQUIREMENTS	Properties of materials	1. Boiling and melting	points	2. Electrical conductivity	3. Heat conductivity					
TIME ALLOCATION	Weeks 1 – 2	1,5 weeks								

If the Term 2 teaching time is reduced, ensure that learners have a thorough understanding of the following key concepts:

Properties of materials

Grade 7 Term 2 Natural Sciences - Planner & Tracker for Recovery ATP

- Material has physical properties like strength, flexibility, melting and boiling points, electrical and heat conductivity.
- Suitability of material for use depends on their physical properties.
- Water can be a solid (ice), a liquid or a gas (water vapour).
- Melting occurs when a solid substance changes into a liquid state. Melting point is the temperature at which a solid becomes a liquid.
- Different substances have different melting points. Ice melts at about 0°C, Iron melts at 1538°C.
- If you heat a liquid, it will eventually boil. This is called the boiling point.
- Different substances boil at different temperatures. Ice boils at about 100°C, Iron boils at 2862°C.
- Boiling and melting points can vary slightly depending on air pressure or impurities in the liquid.
- Electricity moves as an electrical current. Materials that allow electricity to move through them are called electrical conductors. E.g. metals like copper and steel
- Materials that do not allow electricity to move through them are called insulators. E.g. plastic
- Electrical wiring is made up of copper as it conducts electricity, covered with an insulator like plastic to protect from electrical shocks.

TIME	DBE RECOVERY ATP REQUIREMENTS	NOTES	NECT LESSON PLANS: LESSONS	APPR TEXTE	OVED	DATE COMPLETED
Week 3	Solutions as special	This topic	<u>Grade 6 Term 2 Lesson Plans</u>	S&M Gr6	65 – 74	
1 week	mixtures 1 Solutions	nas been recovered	Lesson 2B: Mixtures Lesson 34: Solutions with uniform	VIVA Gr6	63 – 72	
	2. Soluble substances	from Grade 6. Term 2.	appearance	PLAT Gr6	70 – 80	
	3. Insoluble		Lesson 3B: Separating solutions	SFA Gr6	107 – 119	
	substances		Lesson 3C: Crystallisation Lesson 4C: Insoluble substances	DbD Gr6	74 – 83	
				OX Gr6	56 – 65	
				SO Gr6	36 – 39	
				TC Gr6	58 – 64	
				SIBB Gr6	22 – 23; 134 - 150	
If the Term 2 teach Solutions as spec	ing time is reduced, ensure cial mixtures	e that learners he	ave a thorough understanding of the follow	ing key conce	epts:	
		in a si sejin da d	line odłana dru binnije ban bilane da sunku			

- Solutions are a special type of mixture. A solution is a mixture of a solid and a liquid when the solid looks like it has dissolved into the liquid.
- A solution is uniform when it looks the same right through we cannot see the different parts of the mixture.
- Solids can be soluble or insoluble. Soluble a solid that dissolves in water. Insoluble a solid that does not dissolve in water. Give examples of each.
- The solids that dissolve are called the solutes. The liquid that the solute dissolves in is called the solvent.
- Soluble substances: some can be recovered by evaporation (heating) or dissolving. E.g. Salt separated from sea water
- Crystallisation is the process of forming crystals from a solution, through heating the liquid so that it evaporates (changes to a gas) and leaves the solute behind as crystals
 - In a solution, the solute (e.g. Salt) seems to disappear into the solvent (e.g. Water). The solute particles are spread between the solvent particles. This is called dissolving.
- Insoluble substances: some substances will not form a solution in water. They do not dissolve in water. E.g. Oil and water

DATE COMPLETED	80	80	87	143	06	69	41	73	169	-
OVED	75-8	73 - 8	81 -	127 – 1	86 -	66 – (40 - 1	- 99	156 - 1	
APPR TEXTE	S&M Gr6	VIVA Gr6	PLAT Gr6	SFA Gr6	DbD Gr6	OX Gr6	SO Gr6	TC Gr6	SIBB Gr6	-
NECT LESSON PLANS: LESSONS	<u>Grade 6 Term 2 Lesson Plans</u>	Lesson 5A: Factors that affect	uissoiviiig Lesson 5B: Investigating rates of	dissolving						
NOTES	This topic	rias peen recovered	from Grade 6, Term 2.							
DBE RECOVERY ATP REQUIREMENTS	Dissolving	1. Rates of dissolving								
TIME	Week 4	1 week								

If the Term 2 teaching time is reduced, ensure that learners have a thorough understanding of the following key concepts: **Dissolving**

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Temperature, stirring, shaking, and size of the solute all affect the rate of dissolving.

Opportunities to experiment and see the effects of stirring, shaking and size of the solute on how fast the solute dissolves. •

TIME	DBE RECOVERY ATP REQUIREMENTS	NOTES	NECT LESSON PLANS: LESSONS	APPROV TEXTBO	/ED OKS	DATE COMPLETED
Week 5 and 6	Separating mixtures		Grade 7 Term 2 Lesson Plans	OX Gr7	64 – 75	
2 week			Lesson 3A: What are mixtures?	VA Gr7	68 – 75	
	 NIXtures Methods of physical 		magnets	PLAT Gr7	80 - 93	
	separation		Lesson 3C: Filtration and evaporation	SO Gr7	70 – 81	
	3. Sorting and recycling materials		Lesson 4A: Distillation and chromatography	SbS Gr7	82 – 96	
			Lesson 4B: Practical task: Separating a	PEL Gr7	98 – 112	
			mixture	SFA Gr7 1	30 – 163	
			Lesson 4C: Sorting and recycling materials	TC Gr7	84 – 95	
				SIBB Gr7 1	75 - 199	
If the Term 2 teach	ing time is reduced, ensure	that learners h	ave a thorough understanding of the followi	ng key concept	is:	
Separating mixtui	<u>es</u>					
 Define pure s substance. 	ubstances as being made u	ıp of only 1 type	e of particle. Explain that a mixture is made	up two or more	substance	s – impure
Explain the di	fference between mixtures	that can be sep	barated by physical means and those that c	annot.		
Sieving is use	ed to separate mixtures whe	ere the solids ha	ave different sizes. E.G. Separating sand ar	L.G. JULING II U Id stones	III NÀ SIZE A	וום קטמוונץ.
 Magnets will a mixture. E.G. 	attract metals like iron, stee Removing metals from was	l and nickel. In ste dumps.	a mixture, if one of the substances is magn	etic, a magnet (can be use	d to separate the
 Define the ter 	ms solution, solute, solvent	and insoluble.				
Filtration is a	method to separate an inso	luble solid from	ו a liquid. E.G. Separating sand and water t	hrough materia	l like filter p	aper or cloth.
Distillation inv to extract pure	olves evaporation of the so e water from a saltwater so	lvent, followed lution.	by condensation and collection of the cond	ensed liquid. Th	ne lieberg c	ondenser is used
Chromatogra	phy is used to separate mix	tures of differer	nt pigments/colours, through adding anothe	r solvent like wa	ater or alco	hol.
The importan and space.	ce of re-using and recycling	l materials to re	duce waste to avoid pollution and spread o	f disease and t	o avoid wa	ste of materials

Tracker: Grade 7 Term 2

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TIME	DBE RECOVERY ATP REQUIREMENTS	NOTES	NECT LESSON PLANS: LESSONS	APPR(TEXTB	OVED OOKS	DATE COMPLETED
Week 7 and 8	Acids, bases and		<u>Grade 7 Term 2 Lesson Plans</u>	OX Gr7	76 – 81	
2 weeks	neutrals 1 Tastes of		Lesson 5A: What does it taste like?	VA Gr7	76 – 81	
	substances		Lesson 5D: Identifying actus Lesson 5C: Identifying bases	PLAT Gr7	95 – 108	
	2. Properties of acids,		Lesson 6A: Identifying neutrals	SO Gr7	83 – 90	
	bases and neutrals 3 Acid-base indicators		Lesson 6B: Litmus paper tests	SbS Gr7	97 – 104	
				PEL Gr7	118 – 129	
				SFA Gr7	168 – 177	
				TC Gr7	96 – 102	
				SIBB Gr7	206 - 231	
If the Term 2 teach Acids. bases and	hing time is reduced, ensur I neutrals	e that learners h	nave a thorough understanding of the follov	wing key conc	epts:	

The human tongue has taste buds that allow us to identify the 4 different tastes – salty, sweet, sour and bitter.

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- Acids taste sour, they feel rough on the skin. Many acids are strong, corrosive and dangerous to taste and feel. E.g. household cleaning products.
- Some foods contain acids, like lemons.
- Bases taste bitter and feel slippery on the skin. Many bases are caustic and dangerous to taste and feel. Bases are the opposite of acids. If the base is soluble in water it is known as an alkali. Bicarbonate of soda is an example of a base.
- Neutral substances are neither bases nor acids. They are not dangerous. Water, cooking oil, sugar solutions and salt solutions are examples of neutrals.
- A neutral substance can be made by reacting an acid with a base. pH scale is used to measure the strength of acids (pH0 to just under pH7) or a base (just above pH7 to pH14). A neutral is in the middle and has a pH of 7.
- Litmus paper is an indicator that changes colour when it comes into contact with an acid, base or a neutral. The substance that is being tested must be in liquid form.

Below is a sample assessment test and memorandum. Please feel free to use this task as is, or to adapt for your context. It is important to ensure that learners are only assessed on work that has been taught.

Natural Sciences Grade 7 Term 2 Practical Task 20 Marks

Time allocation: 60 minutes (20 minutes preparation, 40 minutes task time)

NOTES TO THE TEACHER

- 1. This practical activity will be completed as part of Section E of lesson 4B.
- 2. This practical will take place during the lesson after the teaching component in Section D, "Accessing Information".
- 3. The first 20 minutes will be used to teach Section D and prepare learners for the practical task.
- 4. The next 40 minutes will be used to complete the practical activity as outlined in Section E.
- 5. The instructions and content of the practical task should be written on the chalkboard for the learners.
- 6. This task will be done in groups of 6.
- 7. Each group will need the following in order to complete the investigation:
 - Two glass jars
 - a container of water
 - a tablespoon of sand
 - a teaspoon of salt
 - a tablespoon of a mixture of samp and beans (or a few small stones and leaves)
 - a piece of filter paper, cloth or paper towel
 - a spoon or stick for stirring
 - a plate, polystyrene tray or piece of newspaper to work on
 - a funnel (this can be constructed out of a few sheets of paper or the top of a coke bottle can be cut off)



- 8. Ensure that you have all the materials ready and prepared for the learners before the lesson begins.
- 9. The memorandum for assessing the practical task is provided.
- 10. The learners should complete the drawings with a sharp pencil and the written answers should be completed in pen.

This activity will be done in groups.

- 1. To do this activity, each group will need the following:
 - Two glass jars
 - a container of water
 - a tablespoon of sand
 - a teaspoon of salt
 - a tablespoon of a mixture of samp and beans (or a few small stones and leaves)
 - a piece of filter paper, cloth or paper towel
 - A spoon or stick for stirring
 - a plate, polystyrene tray or piece of newspaper to work on
 - a funnel (this can be constructed out of a few sheets of paper or the top of a coke bottle can be cut off)



- 2. Ensure you have these materials prepared for each group before the lesson starts.
- 3. Tell the learners that they are going to be doing an investigation where they will be looking at separating mixtures.
- 4. They will be looking at a mixture made up of 4 solids and a liquid.
- 5. They will be planning and conducting an investigation on how to separate this mixture
- 6. Divide the learners into groups of six.
- 7. Write the following onto the chalkboard (always try to do this before the lesson starts):

PRACTICAL TASK

- 1. This practical task will be done in groups of 6.
- 2. Each group will be doing tasks to separate a mixture.
- 3. Each person in the group must participate in the investigation and complete the answers to the written activities in their workbooks.
- 4. Each group will need the following materials and equipment to do the investigation:
 - Two glass jars
 - a container of water
 - a tablespoon of sand
 - a teaspoon of salt
 - a tablespoon of a mixture of samp and beans (or a few small stones and leaves)
 - a piece of filter paper, cloth or paper towel
 - a spoon or stick for stirring
 - a plate, polystyrene tray or piece of newspaper to work on
 - a funnel
- 8. Read through the practical task with the learners.
- 9. Remind the learners that in previous lessons they learnt about a number of different ways to separate mixtures.
- 10. Remind them that they have looked at hand-sorting, filtration, evaporation, seiving and using magnets.
- 11. Tell the learners that today they are going to be conducting a practical task where they are going to be separating a mixture.
- 12. Have each group collect the equipment they will need (as listed on the board) for the task.
- 13. Write the following 'Investigation method' onto the chalkboard:

METHOD

- 1a. Mix the sand, salt, samp and beans (or leaves and stones) together in one of the glass jars.
- 1b. Now pour enough water into the same jar to almost fill it.
- 1c. Give the mixture a good stir.
- 15. Read through the task with the learners.
- 16. Ask them if they have any questions.
- 17. Tell the learners they have 5 minutes to complete this task.
- 18. Supervise the learners whilst they complete the task and answer any questions that they may have.
- 19. After 5 minutes call the learners back to attention.
- 20. Tell the learners that they are now going to work together as a group to separate the mixture into water, salt, sand and samp and beans (or leaves and stones).
- 21. The following will need to be written on the chalkboard:

<u> Task 1:</u>

(13 marks)

- 1a. Choose a name for this practical task.
- 1b. Describe how you would separate the solids from the liquids.
- 1c. What is this separation method called?

As a group, do the actual task of separation using your chosen method.

- 1d. Were you able to separate all of the solids from the liquid?
- 1e. What solids were you able to separate from the liquid?
- 1f. Which solids were you unable to separate from the liquid using this method?
- 1g. Why were you unable to separate the salt from the water using this method?
- 1h. What method could you use to separate the salt and dissolved sand from the water?
- 1i. Explain how you would carry out this method of separation.
- 1j. Once you are left with the sand and salt, what method could you use to separate them?
- 22. Read through task 1 with the learners.
- 23. Ask them if they have any questions.
- 24. Tell the learners they have 10 minutes to complete task 1 and to answer the questions in their workbooks.
- 25. Supervise the learners whilst they complete the task and answer any questions they may have.
- 26. After 15 minutes call the learners back to attention.
- 27. Tell the learners that they are now going to work together, as a group, to separate the solids that have been separated from the liquid.
- 28. The following will need to be written on the chalkboard:

<u>Task 2:</u>

(7 marks)

Pour the solids that you were able to separate from the liquid onto the plate or piece of newspaper.

- 2a. What method can you use to you separate the samp and beans (or stones and leaves) from the wet sand?
- 2b. What method can you use to separate the samp and beans (or stones and leaves) from each other?

Try and separate the larger solids (samp and beans or stones and leaves) from the wet sand.

- 2c. Are the larger solids (samp, beans, stones, leaves) completely free of the sand?
- 2d. Why are the larger particles not completely free from the sand?
- 2e. How can this sand be removed from the larger solids (samp, beans, stones, leaves)?

- 29. Read through task 2 with the learners.
- 30. Ask them if they have any questions.
- 31. Tell the learners they have 10 minutes to complete task 2 and to answer the questions.
- 32. Supervise the learners whilst they complete the task and answer any questions they may have.
- 33. After 10 minutes call the learners back to attention.
- 34. Have learners hand in their workbooks for assessment.
- 35. Learners must then tidy up investigation areas and hand back equipment.

Natural Sciences Grade 7 Term 2 Practical task Memorandum 20 Marks

Торіс	Activity	Expected answer/outcome	Marks
	1		
Separating mixtures	1a.	Separating a mixture of solids and liquids√	1
Separating mixtures	1b.	The funnel was lined with a filter paper (or paper towel) and the mixture is poured through the filter allowing the solids to collect in the top and the liquid to pour through. $\checkmark \checkmark$	2
Separating mixtures	1c.	Filtration✓	1
Separating mixtures	1d.	No✓	1
Separating mixtures	1e.	The samp and beans (stones and leaves)✓ Some of the sand✓	2
Separating mixtures	1f.	The salt \checkmark Some of the sand \checkmark	2
Separating mixtures	1g.	The salt has dissolved into the water \checkmark	1
Separating mixtures	1h.	Evaporation ✓	1
Separating mixtures	1i.	Place the jar in the sun and wait for the water to evaporate off \checkmark	1
Separating mixtures	1j.	Sieving✓	1
	2		
Separating mixtures	2а.	Hand sorting√√	2
Separating mixtures	2b.	Hand sorting√√	2
Separating mixtures	2c.	No✓	1
Separating mixtures	2d.	Wet sand is still stuck to the larger objects✓	1
Separating mixtures	2e.	Wait for the sand to dry and then dust it from the larger objects✓	1
			TOTAL: 20

Natural Sciences Grade 7 Term 2 Test 80 Marks 70 Minutes

NOTE TO THE TEACHER:

If possible, photocopy this test for each learner. If this is not possible, write the test on the chalkboard.

INSTRUCTIONS TO THE LEARNERS

- 1. Answer all questions in blue or black ink.
- 2. Read each question carefully before answering it.
- 3. Pay attention to the mark allocations.
- 4. Plan your time carefully.
- 5. Write your answers in the spaces provided.
- 6. Write neatly.

Practice Question

Read the question and circle the letter that shows the correct answer.

What do we call the process when plants make their own food ?

- A. hydrosphere
- B. lithosphere
- C. atmosphere
- D. nanosphere

You have answered correctly if you have circled (B)

Natural Sciences Grade 7 Term 2 Test 80 Marks

PART 1: Life and Living	
Question 1: Multiple choice	[4]
Read each question and circle the letter that shows the correct answer.	
1a. Which one of these is <u>NOT</u> part of the atmosphere?	(1)
A. Nitrogen.	
B. Water vapour.	
C. Carbon dioxide.	
D. Soil	
1b. Which of these statements is FALSE?	<u>(</u> 1)
A. The hydrosphere is the parts of the Earth covered in water.	
B. Dead organic matter was once living matter.	
C. Water is essential for all life on Earth.	
D. Plants are only found on the lithosphere.	
1c. Which of these statements is <u>TRUE</u> ?	(1)
A. We call the living part of the Earth, the organic part.	
B. We call the non-living part of the Earth, the biotic part.	
C. We call the living part of the Earth, the inorganic part.	
D. We call the non-living part of the Earth, the botanical part.	
1d. Which one of these is not one of the 5 classification kingdoms?	(1)
A. Bacteria.	
B. Fungi.	
C. Organisms.	
D. Protista.	

Question 2: Match the columns [4] Instructions: Match the sentences in COLUMN A with the words in COLUMN B. Draw a line to join the sentence in COLUMN A with the correct word in • COLUMN B. Do this as shown in the example below. **COLUMN A COLUMN B** Needed by all living things example A. Chlorophyll to survive Used by plants to get water 2a. B. Roots and nutrients from soil Allows plants to make their 2b. own food using energy C. Carnivores from the sun 2c. Species: Homo sapiens D. Humans Animals that eat other 2d. E. Air animals **Question 3** [6] Complete the following sentences using words in the block below: gills, cartilage, fins, cold-blooded, eggs, shells, scales

Rewrite the sentences and underline your answers.

3a. Fish have a moist skin covered in _____.

3b. Fish have a skeleton that is made of hard bone or _____.

3c. Fish have ______ which are used for swimming.

3d. Fish have ______ which are used to take oxygen out of the water for breathing.

3e. To reproduce, fish lay ______ that have no ______.

3f. Fish are ______ which means their body temperature changes with the temperature of the environment.

Question 4 [5] Write the word that is being described in the sentence. Only write the answer. 4a. Animal with a backbone. 4b. Phyla in invertebrate group with soft bodies and a shell for protection. 4c. Phyla of most common animals on Earth. 4d. Animals without a backbone. 4e. Animal class to which snakes belong. Question 5 [2] "Plants with seeds are divided into Angiosperms and Gymnosperms Explain the main differences in reproduction between angiosperms and gymnosperms.

[4]

Question 6

Complete the following table showing the differences between plants and animals:

FLOWER PART	FUNCTION
Anther	
Stigma	
Ovary	
Petals	

Question 7	[5]
Read the following statements and say whether each one is true or false:	
7a. Seeds can be dispersed by wind	
7b. Cross-pollination is when pollen is transferred from the anther of one flower	
to the stigma of another flower.	
7c. HIV can be spread by kissing	
7d. If I use contraception, a pregnancy can never happen	
7e. There is no cure for HIV infection.	

Question 8	[10]
"Puberty is a confusing time. Our bodies are changing and we often don't feel in control of our emotions."	
8a. Using what you have learnt, and the words in the box below, write 5-7 lines explaining wh you understand by puberty and what happens to a girl's and a boy's body during puberty.	ıat
ages, hair, skin, breasts, menstruation, hips, voices, penis, hormones, sweating, feelings,pregnancy, sexuality	
8b. Name the hormone that causes these changes in girls.	
8c Name the hormone that causes these changes in boys.	

PART 2: Matter and Materials	
Question 9: Multiple choice	[4]
Read each question and circle the letter that shows the correct answer.	
9a. Which one of these is <u>NOT</u> a physical property of materials?	(1)
A. Strength.	
B. Flexibility.	
CCost	
D. Conductivity	
9b. Which of these statements is <u>FALSE?</u>	<u>(</u> 1)
A. Matter takes up space and has mass.	
B. Matter can be a liquid, solid or gas.	
C. Flexible materials break easily.	
D. Materials can be natural or man-made.	
9c. Which of these statements is <u>TRUE</u> ?	(1)
A. All materials have the same boiling point.	
B. Ice melts at around 0°C.	
C. Water in Johannesburg and water in Cape Town boils at the same temperature.	
D. The boiling point is the same as the melting point of a material.	
9d. Which one of these materials is a good conductor of heat?	(1)
A. Metal.	
B. Wood.	
C. Plastic.	
D. Rubber.	

Question 10: Match the columns [4] Instructions: • Match the sentences in COLUMN A with the words in COLUMN B. Draw a line to join the sentence in COLUMN A with the correct word in • COLUMN B. Do this as shown in the example below. **COLUMN A COLUMN B** Related to or producing example A. Rigid heat Unable to bend or be 10a. B. Insulator forced out of shape Can be broken down by 10b. C. Biodegradable bacteria A material that does not 10c. D. Tensile allow the flow of electricity The strength of an object 10d. that stops it from breaking E. Thermal when pulled **Question 11** [3] Complete the following sentences using words in the block below: variables, fair test, constant, conclusions,

Rewrite the sentences and underline your answers.

11a. A ______ is a scientific investigation that is carefully controlled.

In a fair test, we compare two or more things.

11b. The things that remain the same in a fair test are called the ______.

11c. The things we change in a fair test are called the _____.

Question 12:	[5]
Write the word that is being described in the sentence.	
Only write the answer.	
12a. A mixture made up of a solid that is dissolved into a liquid.	
12b. The property of a solid that does not dissolve into a liquid.	
12c. Small, microscopic parts that make up matter.	
12d. The change of state from gas to liquid.	
12e. A method of separating two liquids that have different boiling points.	
Question 13:	[2]
	[4]
Solute + solvent = solution	
Describe a solute and a solvent, highlighting what is different between the two.	
13a. Solute:	
13b. Solvent:	

Question 14				[6]
Place the followin	ng substances under t	the correct columns:		
	washing powder, oil,	, water, lemon, bicarbonate	of soda, vinegar	
Ē		NEUTRAL	BASES	
Question 15				[5]
Read the followi	ng statements and sa	ay whether each one is true	or false:	
15a. Indicators	s are dyes that chang	e colour in acids and bases		
15b. Acids car				
	is found in stomach	juices		
15d. Bases are	e the opposite of acid	Is and can never be danger	ous	
15e. Bases ca	n be caustic and can	burn organic tissue.		
Question 16				[4]
Rewrite the senter	nces filling in the miss	ing words from the list belo	w.	
Underline the word	ds you fill in.			
	solids matter regula		, particles, move	
		ar, big, small, liquids, gases		
All matter is made	up of particles.	ar, big, small, liquids, gases		
All matter is made 16a. In	up of particles.	ar, big, small, liquids, gases	pattern.	
All matter is made 16a. In 16b. In	up of particles. , the particle	ar, big, small, liquids, gases es are closely packed in a _ paces between the	pattern.	
All matter is made 16a. In 16b. In	up of particles. , the particle , there are s around each_c	ar, big, small, liquids, gases as are closely packed in a paces between the other.	pattern.	
All matter is made 16a. In 16b. In 16c. In	e up of particles. , the particle , there are s around each c , the particle	ar, big, small, liquids, gases as are closely packed in a paces between the other. as move in all directions.	pattern.	

Question 17:

Three factors affect the rate of a solute dissolving in a solvent.

- 1. Stirring
- 2. _____
- 3. _____

Question 18:

Complete the table below

TABLE OF MIXTURES

MIXTURE	Is the mixture a solution after stirring?	Which substance is the solvent?	Which substance is the solute?
oil and water	No	N/A	N/A
vinegar and water	a)	water	vinegar
sand and water	b)	N/A	N/A
sugar and water	c)	d)	e)
			TOTAL: 80

[2]

[5]

Natural Sciences Grade 7 Term 2 Test Memorandum 80 Marks

PART 1: Life and Living

CAPS Topic	Questions	Expected answer(s)	Marks
	1		
The biosphere	1a.	D✓	1
The biosphere	1b.	D✓	1
The biosphere	1c.	A✓	1
Biodiversity	1d.	C✓	1
	2		
Biodiversity	2a.	B✓	1
Biodiversity	2b.	A✓	1
Biodiversity	2c.	D✓	1
Biodiversity	2d.	C✓	1
	3		
Biodiversity	За.	Scales✓	1
Biodiversity	3b.	Cartilage✓	1
Biodiversity	Зс.	Fins√	1
Biodiversity	3d.	Gills√	1
Biodiversity	30	eggs√	2
	JC.	shells√	
Biodiversity	3f.	cold-blooded√	1

	4			
Biodiversity	4a.	Vertebrate✓		1
Biodiversity	4b.	Molluscs√		1
Biodiversity	4c.	Arthropods✓		1
Biodiversity	4d.	Invertebrates✓		1
Biodiversity	4e.	Reptiles√		1
	5			
Sexual reproduction	5.	Angiosperms reproduce by producing seeds in flowers. Gymnosperms produce their seeds in cones $\checkmark \checkmark$		2
	6			
Biodiversity		FLOWER PART	FUNCTION	4
		Anther	Makes the pollen grains that hold the male sex cells√	
		Stigma	The sticky tip of the female part that receives the pollen√	
		Ovary	The ovary holds the ovules which are the female sex cells√	
		Petals	Attract pollinators like insects and birds√	
	7			
Sexual reproduction	7a.	True√		1
Sexual reproduction	7b.	True√		1
Sexual reproduction	7c.	False√		1
Sexual reproduction	7d.	False√		1
Sexual reproduction	7e.	True√		1

	8		
Sexual reproduction	8a.	(Any 8 in total) 🗸 🗸 🗸 🗸 🗸 🗸	8
		In girls:	
		Can occur from the age of 10	
		Breasts develop	
		Hair grows on the vagina	
		Hair grows under arms	
		Skin becomes oily	
		Hips widen	
		Menstruation begins	
		Feelings can become confusing	
		In boys:	
		Hair grows around penis	
		and scrotum	
		Hair grows under arms	
		Facial hair starts to grow	
		Skin becomes oily	
		Voice becomes deeper	
		Penis becomes bigger	
		Sperm is produced	
		Increase in sweating	
Sexual reproduction	8b.	Oestrogen	1
Sexual reproduction	8c.	Testosterone	1

PART 2 : Matter and Materia	ls		
	9		
Properties of materials	9a.	C✓	1
Properties of materials	9b.	C✓	1
Properties of materials	9c.	В√	1
Properties of materials	9d.	A✓	1
	10		
Properties of materials	10a.	A✓	1
Properties of materials	10b.	C✓	1
Properties of materials	10c.	В✓	1
Properties of materials	10d.	D✓	1
	11		
Separating mixtures	11a.	fair test ✓	1
Separating mixtures	11b.	constant ✓	1
Separating mixtures	11c.	variables ✓	1
	12		
Separating mixtures	12a.	solution 🗸	1
Separating mixtures	12b.	insoluble ✓	1
Separating mixtures	12c.	particles ✓	1
Separating mixtures	12d.	condensation ✓	1
Separating mixtures	12e.	distillation \checkmark	1
	13		
Separating mixtures	13a.	Solute: The substance that dissolves when making a solution e.g.: salt in a salt water solution ✓	1
Separating mixtures	13b.	Solvent: The liquid in which the solute dissolves e.g.: water in a salt water solution \checkmark	1

	14				
Acids, bases and neutrals	14	ACID	NEUTRAL	BASE	6
		lemon	oil	washing powder	
		vinegar	water	bicarbonate of soda	
	15				
Acids, bases and neutrals	15a.	True ✓			1
Acids, bases and neutrals	15b.	True ✓			1
Acids, bases and neutrals	15c.	False ✓			1
Acids, bases and neutrals	15d.	False ✓			1
Acids, bases and neutrals	15e.	True ✓			1
	16				
Solids, liquids and gases	16a.	solids ✓ regu	ulart ✓	1/2 mark each	
Solids, liquids and gases	16b.	liquids ✓ partic	les ✓ move ✓	1/2 mark each	8 x
Solids, liquids and gases	16c.	gases √		1/2 mark each	= 4
Solids, liquids and gases	16d.	liquids ✓ sma	all ✓	1/2 mark each	
	17				
Dissolving	17a.	Heating ✓			1
Dissolving	17b.	Grain size ✓			1
	18				
Solutions as special mixtures	18a.	Yes ✓			1
Solutions as special mixtures	18b.	No ✓			1
Solutions as special mixtures	18c.	Yes ✓			1
Solutions as special mixtures	18e.	Water ✓			1
Solutions as special mixtures	18f.	sugar			1
				тот	AL: 80